

are unlikely to be successful. The parameter  $t/y_c$  must be included in the correlation equation.

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## THERMAL CONDUCTIVITY OF PACKED BEDS AND POWDER BEDS

S. C. CHENG

Mechanical Engineering Department, University of Ottawa, Ottawa 2, Canada

and

R. I. VACHON

Department of Mechanical Engineering, Auburn University, Alabama 36830

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## NOMENCLATURE

$A_{com}$	area of continuous phase perpendicular to the $xy$ plane;	$p$	pressure of two particles against each other;
$\bar{a}$	accommodation coefficient;	$q$	integer defined in equation (20);
$B$	constant in equation (1);	$R$	thermal resistance;
$C$	constant in equation (1);	$R'$	thermal resistance in the microgap;
$D$	constant in equation (16);	$R_L$	thermal resistance determined by narrowing of the lines of the heat flow in the region adjacent to the place of contact;
$E$	Young's modulus;	$R_o$	thermal resistance of the oxidic film;
$H$	gas pressure in the pore;	$R_{sp}$	thermal resistance of microroughness at the place of contact;
$H_o$	normal gas pressure, ( $9.93 \times 10^4 N/m^2$ );	$r_{sp}$	radius of a contact spot;
$h$	thickness and width of the skeleton of Fig. 1(d);	$T$	mean absolute temperature of packed beds or powder beds.
$h_r$	height of particle microroughnesses;		
$k$	ratio of gas heat capacities at constant pressure and volume;		
$k_k$	coefficient of particle adhesion;		
$k_m, k_n$	empirical coefficients, $k_m = (h_r/L) \times 10^3$ ;		
$L$	characteristic size of an elementary cell (particle diameter);		
$l$	characteristic size of a pore;		
$n$	number of particles in a layer defined in equation (15);		
$P_d$	discontinuous phase volume fraction;		
$Pr_r$	Prandtl number;		

## Greek letters

$\Delta R$	thermal resistance in a layer;
$\Delta x$	small increment in the $x$ axis;
$\delta$	one-half of $h$ ;
$\eta$	constant defined in equation (7);
$\lambda_o$	gas molecular free path at normal pressure;
$\lambda$	thermal conductivity;
$\lambda'$	thermal conductivity in the microgap;
$\lambda_o$	thermal conductivity of gas at normal pressure;

- $\mu$ , Poisson's ratio;
- $v_g$ , ratio of  $\lambda_g/\lambda_s$ ;
- $v'_g$ , ratio of  $\lambda'_g/\lambda_s$ .

Subscripts

- c*, contact;
- e*, equivalent;
- ec*, equivalent continuous phase;
- g*, gas;
- gm*, gas conduction;
- gr*, gas radiation;
- s*, solid.

INTRODUCTION

THE TECHNIQUE reported by the writers for predicting the thermal conductivity of heterogeneous solid mixtures [1] is modified to account for contact resistance. The contact

resistance model discussed in reference [2] is combined with the technique. Results of applying the modified technique to powder and packed beds are compared with experimental data. Other studies [3-18] have been considered in preparing this communication.

ANALYSIS

A schematic of a packed bed or powder bed of unit volume is shown in Fig. 1(a). Based on the development in [1], the discontinuous phase, mainly gases in the pores of the packed beds or powder beds, can be rearranged in the continuous phase, mainly particles, and expressed by a parabolic distribution;

$$y = B + Cx^2 \tag{1}$$

where  $B = \sqrt{(3P_d/2)}$  and  $C = -4\sqrt{(2/3)P_d}$ , as shown in Fig. 1(b).

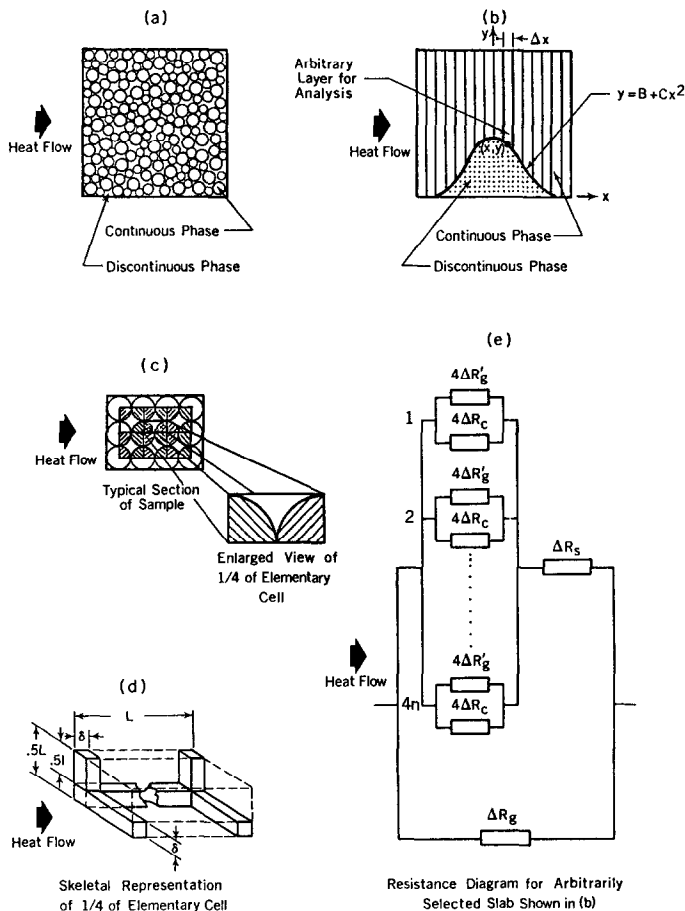


FIG. 1. Model for the study of the thermal conductivity of packed beds and powder beds.

The contact resistance of the particles in the continuous phase can be calculated as follows [2]. Assume the granular particles to be spheres in a cubic arrangement. The elementary cells are shown in Fig. 1(c). Assume the particles to be symmetric. Then, one quarter of an elementary cell, as seen in Fig. 1(c), can be represented in skeletal form as shown in Fig. 1(d). Considering Fig. 1(d), it can be seen that there exists a functional relationship between  $P_d$ , the discontinuous phase volume fraction, and  $h/l$ , the ratio of the thickness or width of the skeleton to the characteristic size of the pore. This relationship is

$$P_d = f(h/l); \tag{2}$$

where  $h = 2\delta$ , the thickness and width of the skeleton in Fig. 1(d) and  $l = L - h$ . The functional relationship plotted

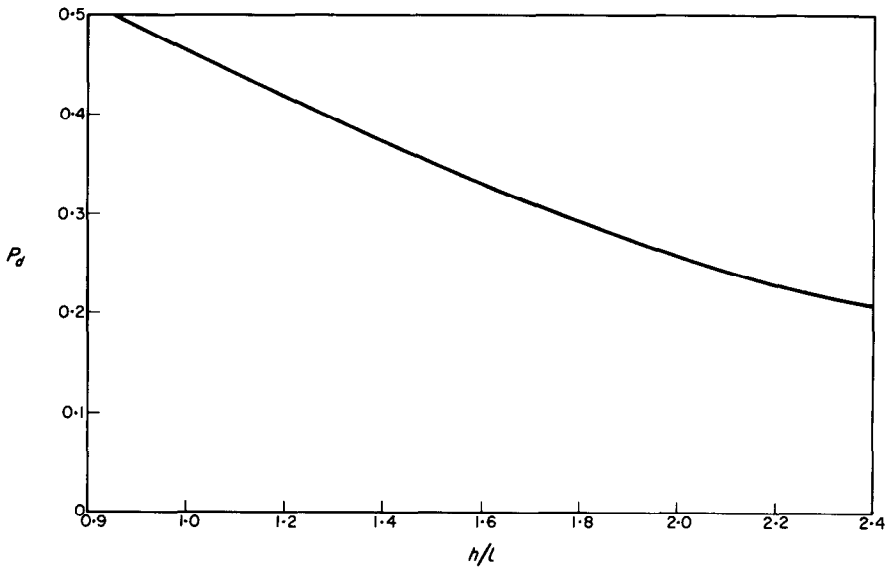


FIG. 2. Graphical representation of  $P_d$  as a function of  $h/l$ .

resistance in an elementary cell. The contact resistance and the resistance of the gas in the microgap for one quarter of the elementary cell can be designated as  $R_c$  and  $R'_g$ , respectively.  $R_c$  and  $R'_g$  can be expressed as

$$R_c = R_L + R_{sp} + R_0 \tag{3}$$

and

$$R'_g = \frac{4h_r k_k}{\lambda'_g h^2} \tag{4}$$

$R_c$ , the contact thermal resistance, is defined as

$$R_c = \frac{1}{\lambda_c L} \tag{5}$$

$R_L$ ,  $R_{sp}$ , and  $R_0$  in equation (3) will be discussed as follows.  $R_L$ , which is the resistance due to the narrowing of the lines

of the heat flow in the region adjacent to the place of contact, can be expressed as

$$R_L = \frac{1}{2r_{sp} \lambda_s}; \tag{6}$$

where

$$r_{sp} = 0.725 \sqrt[3]{(\eta p L / 2)} \tag{7}$$

and

$$\eta = \frac{2(1 - \mu^2)}{E}$$

Equation (7) is Hertz formula [19].  $R_{sp}$ , which is the resist-

ance expressed as

$$R_{sp} = \frac{h_r k_k}{\pi r_{sp}^2 \lambda_s} \tag{8}$$

$h_r$  in equation (8) is the height of particle microroughnesses. The ratio

$$\frac{h_r}{L} = k_m \cdot 10^{-3} \tag{9}$$

is fairly stable for various sizes of particles.  $k_m$  is an empirical coefficient. Equation (8) can be rewritten as

$$R_{sp} = \frac{L k_m k_k}{10^3 \pi r_{sp}^2 \lambda_s}; \tag{10}$$

where  $k_k$  is the coefficient of particle adhesion.  $R_0$ , which is

the resistance of the oxidic film, can be neglected in most cases [20] where

$\lambda_g$ , the thermal conductivity of the gas in a pore, consists of two components,  $\lambda_{gr}$  and  $\lambda_{gm}$ , or

$$\lambda_g = \lambda_{gr} + \lambda_{gm} \quad (11)$$

Similarly,  $\lambda'_g$ , the thermal conductivity of gas in the microgap, consists of  $\lambda'_{gr}$  and  $\lambda'_{gm}$ , or

$$\lambda'_g = \lambda'_{gr} + \lambda'_{gm} \quad (12)$$

$\lambda_{gr}$  and  $\lambda'_{gr}$ , the radiant thermal conductivities of the gas in the pore and in the microgap, can be calculated from the technique proposed by Loeb [8] or Chudnovsky [21]. For low temperature,  $\lambda_{gr}$  and  $\lambda'_{gr}$  can be neglected. According to Prasolov [22],  $\lambda_{gm}$  can be expressed as

$$\lambda_{gm} = \frac{\lambda_0}{1 + B'/Hl}; \quad (13)$$

$$B' = \frac{4k}{k+1} \frac{2-\bar{a}}{\bar{a}} Pr^{-1} A_0 H_0.$$

Similarly

$$\lambda'_{gm} = \frac{\lambda_0}{1 + B'/k_n Hl} \quad (14)$$

For simplicity,  $\lambda_g$  can be assumed approximately equal to  $\lambda'_g$  in the calculation.

Now one can proceed to develop an expression for the thermal conductivity of packed beds and powder beds considering contact resistance. Consider an arbitrary layer with thickness  $\Delta x$  as shown in Fig. 1(b). The corresponding resistance diagram can be drawn on Fig. 1(e). The number of particles,  $n$ , in the layer can be calculated as

$$\eta = \frac{A_{con} \Delta x}{\frac{4}{3} \pi \left(\frac{L}{2}\right)^3} \quad (15)$$

Table 1. Comparison of predicted thermal conductivities with experimentally determined conductivities for packed beds and powder beds

Mixture	$\lambda_g^\dagger$	$\lambda_g$	$P_d$	$L(\text{mm})$	$\lambda_{exp.}$	$\lambda_g^\ddagger$	Percentage of deviation	$\lambda_g^\S$	Percentage of deviation
Steel spheres in air [23]	38.4	0.026	0.38	3.18	0.525 [24]	0.525	0	0.529	0.8
Steel spheres in air [7]	45.0	0.0272	0.413	3.2	0.40	0.51	27.5	0.469	17.3
Steel spheres in air [7]	45.0	0.0272	0.406	3.2	0.60	0.54	-10.0	0.489	-18.5
Plumbum shot in air [7]	34.3	0.0273	0.42	1.6	0.418	0.49	17.2	0.433	3.6
Plumbum shot in air [7]	34.3	0.0273	0.433	6.4	0.404	0.473	17.1	0.39	-3.5
MgO in air at 375°K [21]	24.4	0.0318	0.42	0.268	0.433	0.425	-1.9	0.384	-11.3
MgO in air at 502.4°K [21]	27.9	0.0387	0.42	0.268	0.502	0.515	2.6	0.465	-7.4
MgO in air at 572.1°K [5]	22.1	0.045	0.42	0.268	0.552	0.556	0.7	0.528	-4.4
MgO in air at 723°K [5]	16	0.0533	0.42	0.268	0.661	0.67	1.4	0.602	-8.9
MgO in air at 810°K [5]	13	0.056	0.42	0.268	0.666	0.68	2.1	0.617	-7.4
Average percentage of deviation							8.1		8.3

† Units of  $\lambda$ : W/m deg.

‡ Luikov *et al.* [2].

§ Equation (19).

The total number of quarter elementary cells in the layer is  $4n$ .  $\Delta R_{ec}$ , the equivalent resistance of the continuous phase in the layer, can be expressed for the  $4n$  quarter elementary cells as

$$\begin{aligned} \Delta R_{ec} &= \Delta R_c \\ &+ \frac{1}{\left(\frac{1}{4R_c} + \frac{1}{4R'_g}\right) + \left(\frac{1}{4R_c} + \frac{1}{4R'_g}\right) + \dots + \left(\frac{1}{4R_c} + \frac{1}{4R'_g}\right)} \\ &\quad \text{4n terms} \\ &= \frac{\Delta x}{\lambda_s(1-y)} + \frac{1}{\left(\frac{1}{4R_c} + \frac{1}{4R'_g}\right) \frac{4A_{con}\Delta x}{4} \frac{\pi(L)}{3} \left(\frac{L}{2}\right)^3} \\ &= \frac{\Delta x}{\lambda_s(1-y)} + \frac{4R_c R'_g}{R'_g + R_c} \frac{\pi L^3}{24(1-y)\Delta x} \\ &= \frac{\Delta x}{\lambda_s(1-y)} + \frac{D}{(1-y)\Delta x}; \end{aligned} \quad (16)$$

where

$$D = \frac{\pi L^3 R_c R'_g}{6(R'_g + R_c)}$$

$\Delta R_e$ , the equivalent resistance in the layer, can be expressed as

$$\Delta R_e = \frac{\Delta R_g \Delta R_{ec}}{\Delta R_g + \Delta R_{ec}} \quad (17)$$

Substituting equation (16) into equation (17), yields

$$\begin{aligned} \Delta R_e &= \frac{\Delta x \left[ \frac{\Delta x}{\lambda_s(1-y)} + \frac{D}{(1-y)\Delta x} \right]}{\lambda_g y \left[ \frac{\Delta x}{\lambda_s(1-y)} + \frac{D}{(1-y)\Delta x} \right] + \frac{D}{\lambda_g y}} \\ &= \frac{\left[ \frac{\Delta x}{\lambda_s(1-y)} + \frac{D}{(1-y)\Delta x} \right] + \frac{\Delta x}{\lambda_g y}}{\frac{\Delta x^3 + D\lambda_s\Delta x}{\lambda_g y\Delta x^2 + D\lambda_g\lambda_s y + \lambda_s(1-y)\Delta x^2}} \end{aligned} \quad (18)$$

Summing up all  $x$ 's from  $-\frac{1}{2}$  to  $\frac{1}{2}$ , equation (18) becomes

$$\begin{aligned} R_e &= 2 \sum_0^{B/2} \frac{\Delta x^3 + D\lambda_s\Delta x}{\lambda_g y\Delta x^2 + D\lambda_g\lambda_s y + \lambda_s(1-y)\Delta x^2} \\ &+ 2 \sum_{B/2}^{1/2} \left( \frac{\Delta x}{\lambda_s} + \frac{D}{\Delta x} \right) \\ &= 2 \sum_0^{B/2} \frac{\Delta x^3 + D\lambda_s\Delta x}{\lambda_g y\Delta x^2 + D\lambda_g\lambda_s y + \lambda_s(1-y)\Delta x^2} \\ &\quad + \frac{1-B}{\lambda_s} + \frac{4D}{1-B} \end{aligned} \quad (19)$$

The expression for  $y$ , the distribution function, can be obtained from equation (1). The value of  $\Delta x$  should be compatible with the value of  $L$ . Before computing  $\Delta x$ ,  $q$  which is some integer should be determined first.  $q$  is approximated by

$$q \approx \frac{B/2}{L} \quad (20)$$

Once  $q$  is determined,  $\Delta x$  can be determined by

$$\Delta x = \frac{B/2}{q} \quad (21)$$

Mathematically,  $q+1$  represents the number of terms to be summed in equation (19) for the value of  $\Delta x$  determined from equation (21).

Once  $R_e$  is determined from equation (19),  $\lambda_e$  is simply the numerical reciprocal of  $R_e$ .

A sample problem [2, 23]—steel spheres in air—will be used to illustrate the application of equation (19) in predicting the thermal conductivity of packed beds and powder beds.

Reference data:  $L = 3.18 \times 10^{-3}$ ;  $P_d = 0.38$ ;  $\lambda_s = 38.4$ ;  $\lambda_o = 0.026$ ;  $T = 320^\circ\text{K}$ ;  $h/l = 1.34$ ;  $h/L = 0.573$ ;  $H = 10^5$  N/m<sup>2</sup>;  $v_g = v'_g = 0.677 \times 10^{-3}$ ;  $k_m = 3$ ;  $k_k = 1.5$ ;  $\lambda_c = 0.03$ ; experimental equivalent thermal conductivity  $\lambda_e = 0.525$  W/m deg [24].

Solution:  $L = 3.18 \times 10^{-3}$  and  $h/L = 0.573 \Rightarrow h = 0.0182$   $v_g = \lambda_o/\lambda_s \Rightarrow \lambda_g = v_g\lambda_s = 0.000677 \times 38.4 = 0.026$ .

From equation (5),

$$R_c = \frac{1}{\lambda_c L} = \frac{1}{0.03 \times 0.00318} = 10480.$$

From equations (4) and (5),

$$R'_g = \frac{4Lk_mk_k}{\lambda'_g h^2 10^3} = \frac{4 \times 0.00318 \times 3 \times 1.5}{0.026 \times (0.00182)^2 \times 10^3} = 665.$$

From equation (1),  $P_d = 0.38 \Rightarrow B = 0.755$  and  $C = -5.3$ .

From equation (20),  $q \approx \frac{B/2}{L} \Rightarrow q = 119$ .

From equation (21),  $\Delta x = \frac{B/2}{q} = 0.00317$ .

Substituting all the above data into equation (19), yields

$$\lambda_e = 0.529 \text{ W/m deg.}$$

which agrees within 0.8 per cent with the experimental value of  $\lambda_e$ . Table 1 compares thermal conductivities of packed beds and powder beds as predicted by equation (19) and thermal conductivity values as obtained by the technique proposed by Luikov *et al.* [2] with experimental data available from the literature.  $\lambda_c$  and  $\lambda_g$  in Table 1 can be obtained

experimentally [2] or from equations (5), (11) and (13). The average percentage of deviations of the predicted values from the measured values of thermal conductivity in Table 1 are 8.1 per cent using Luikov *et al.* and 8.3 per cent by equation (19). Deviations in some cases are more than 100 per cent, as shown by Luikov *et al.* [2], when contact resistance is not considered.

#### SUMMARY

A general equation for predicting the thermal conductivity of packed beds and powder beds has been presented. The equation was developed in terms of the thermal conductivity of the particles and the gas in the pores, discontinuous phase volume ratio, contact resistance, size of the particles, radiative transfer in the gaseous pores, pore size, contact pressure between particles, and surface properties of the particles. Although spherical particles were assumed in the model, the equation can be applied to other particle shapes as long as the mean diameter of the particles can be determined.

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